

1 **Design, construction and monitoring of pilot systems to evaluate the effect of freeze-**  
2 **thaw cycles on pollutant retention in wetlands**

3 Elisangela Heiderscheidt\*, Uzair Khan, Katharina Kujala, Anna-Kaisa Ronkanen, Heini  
4 Postila

5 Water, Energy and Environmental Engineering, Faculty of Technology, 90014 University  
6 of Oulu, Finland.

7 \*Corresponding author: Water, Energy and Environmental Engineering, Faculty of  
8 Technology, 90014 University of Oulu, Finland. Phone: +358 (0)8 553 4502 e-  
9 mail:elisangela.heiderscheidt@oulu.fi.

10

11 **Abstract:**

12 Substantial knowledge is available on purification processes occurring in treatment  
13 wetlands. However, there is still a lack of understanding regarding pollutant removal  
14 under winter conditions in cold climate regions, especially with respect to the effect of  
15 freeze-thaw cycles. Due the complexity of soil freeze/thaw processes and the variety of  
16 factors affecting pollutant removal, pilot systems can be seen as powerful tools offering  
17 a rare opportunity to observe processes that have a significant impact on year-round  
18 purification. This paper describes the design, construction, monitoring and operation of  
19 two replicate pilot peat-based wetlands subjected to two simulated freeze-thaw cycles.  
20 Undisturbed peat soil and treated process water samples were collected from a full-scale  
21 treatment wetland operating at a mining site in Northern Finland. In general, the design  
22 approach and monitoring methodology developed were successful and the pilot wetlands  
23 functioned well. Fluctuations in removal efficiency of target compounds due to freezing  
24 and thawing conditions were observed. Overall, removal of sulphate and arsenic  
25 decreased during frost periods, while removal of antimony increased. Monitoring data  
26 from the full-scale treatment wetland were used to assess the representativeness of the  
27 results obtained. Comparisons of seasonal variations in pollutant concentrations in  
28 outflow samples from the full-scale wetland and those measured in the pilot wetlands  
29 revealed similar fluctuations in removal efficiency during frost and frost-free periods,  
30 suggesting that the pilot wetlands simulated the real system rather well. Carefully

31 designed pilot systems can thus be valuable tools for assessing the effect of harsh winter  
32 conditions on wetland processes and operation.

33 **Keywords:** natural wetlands, mining water, metals, nitrogen, arctic conditions

## 34 1. Introduction

35 In general, wetlands are well-suited for purification of water from diffuse sources, as they  
36 retard the flow of water and provide a large filtration network with many adsorptive  
37 surfaces on plant roots or soil particles where nutrients, suspended solids (SS) and other  
38 harmful elements (e.g. metals and humic substances) can be retained (Kadlec and  
39 Wallace, 2009; Vymazal, 2011; Heikkinen et al., 2018; Kujala et al., 2019; Khan et al.,  
40 2019a). Treatment wetlands can be established in natural areas (making use of natural  
41 soil, vegetation etc.) or can be constructed as completely separate, purpose-built systems.  
42 In the Northern hemisphere, more than 50% of the exposed land surface is regularly  
43 influenced by seasonal soil frost (Zhang et al., 2004). Soil freezing is known to limit and  
44 control ecosystem functions (Sulkava and Huhta, 2003; Cleavitt et al., 2008), nutrient  
45 availability (Edwards and Cresser, 1992), soil texture (Kadivar and Manahiloh, 2019) and  
46 processes such as nutrient leaching (e.g. Fitzhugh et al., 2003) and gas fluxes (Groffman  
47 et al., 2006). These are key issues to be considered when wetlands are used for water  
48 purification in cold climate regions. Northern wetlands can experience repeated freeze-  
49 thaw cycles (lasting from days to months) every winter. Short freeze-thaw cycles occur  
50 frequently in the beginning of winter, as cold night temperatures lead to freezing of upper  
51 soil layers, while warmer day temperatures lead to thawing. On the other hand, soil can  
52 be frozen for several months during the middle of winter if an insulating snow cover is  
53 thin or absent.

54 Soil frost and other winter-related conditions can have a significant effect on the pollutant  
55 retention efficiency of wetlands. For example, soil frost can change flow pathways within  
56 wetlands (Postila et al., 2015b) and reduce hydraulic retention time significantly  
57 (Ronkanen and Kløve, 2007; Heikkinen et al., 2018). Physical-chemical processes  
58 (adsorption, retention of particles etc.), which are strongly dependent on contact between  
59 water and soil particles and on residence time of water, can decline, affecting pollutant  
60 removal. In addition, low temperatures have a direct impact on microbe-driven processes  
61 (e.g. nitrification, denitrification, sulphate ( $\text{SO}_4^{2-}$ ) reduction) by decreasing process rates,

62 thus leading to reduced removal efficiencies (Smid and Beauchamp, 1976; Knoblauch  
63 and Jørgensen, 1999; Kadlec and Reddy, 2001; Stein et al., 2007; Tourna et al., 2008;  
64 Postila 2015a). Although substantial knowledge is available on wetland processes, the  
65 overall effect of cold climate conditions on water purification efficiency is not entirely  
66 clear. For example, there is still a lack of understanding regarding pollutant removal (e.g.  
67 nitrogen, metals,  $\text{SO}_4^{2-}$ ) under winter conditions, especially for organic soils, and the  
68 effect of freeze-thaw cycles. As wetlands (and similar systems such as buffer zones etc.)  
69 are extensively used in cold climate regions for pollution control at e.g. mining sites, peat  
70 extraction sites and urban areas (Heikkinen et al., 2018; Kujala et al., 2019), there is a  
71 clear need for knowledge and technological advances as regards identifying and  
72 understanding the effect of freeze-thaw cycles on pollutant retention processes.

73 The overall aims of this study were to extend available knowledge on the effect of freeze-  
74 thaw cycles on pollutant retention in wetlands and to develop technological and  
75 methodological solutions for monitoring soil frost and thawing processes in these  
76 systems. The novelty of this work lies in the use of pilot wetland systems designed to  
77 simulate real freeze-thaw conditions in a controlled environment. Due the complexity of  
78 soil freeze/thaw processes and the variety of factors affecting pollutant removal in  
79 wetlands, pilot systems can be seen as powerful tools offering a rare opportunity to  
80 observe processes that have a significant impact on year-round purification.

81 The design, implementation and operation of two pilot wetlands submitted to fully-  
82 controlled freeze and thaw cycles was studied. The objectives were i) to create freezing  
83 and thawing environments that replicated real conditions as accurately as possible and ii)  
84 to monitor parameters that can be used in combination to identify possible pollutant  
85 retention processes. The main research questions addressed were: Is it possible to  
86 construct a cost-effective pilot wetland system where freeze-thaw cycles can be simulated  
87 in a way that satisfactorily replicates real conditions? Do observations made in a pilot  
88 wetland system match observations made in a full-scale treatment peatland?

## 89 **2. Design and construction of pilot systems**

90 This study was conducted a part of a project seeking ways to minimise the environmental  
91 impacts of the mining industry in high-latitude regions (Min-North, Interreg Nord).  
92 Therefore the pilot systems were tested for purification of mining water and peat was

93 selected as the medium in the pilot wetlands. Peat is a type of organic soil abundant in  
94 boreal regions and peatlands are widely used as buffer zones receiving mining-affected  
95 waters prior to discharge to recipient water bodies such as lakes or rivers.

96 Mining-affected water and undisturbed peatland soil samples from a full-scale treatment  
97 wetland purifying treated excess mining process water since 2010 in Northern Finland  
98 were used in pilot wetland construction and operation (two replicate units). Long-term  
99 monitoring data and data on operating parameters (e.g. water residence time, average frost  
100 depth) from the same site were used as the basis for the pilot design. In addition, climate  
101 data from the study site location were used to define the freeze-thaw conditions simulated.

## 102 **2.1 Dimensioning parameters, soil sampling and stabilisation procedure**

### 103 *2.1.1 Pilot wetland sizing*

104 The actual dimensions of the peat samples collected from the treatment wetland were  
105 based on practical factors that were critical for the sampling procedure. Samples were  
106 taken in the beginning of May 2017 during the frost period (i.e. when surface peat layers  
107 were frozen) to avoid collapse of the peat sample at the time of sampling. Due to the  
108 nature of the wetland soil and its position in the landscape, motorised machinery/vehicles  
109 could not be used to assist in sampling. Thus the procedure used was manual, simple and  
110 cost-effective. Readily available plastic boxes (moving boxes) were selected as sample  
111 containers, which allowed for sample dimensions of 50 cm x 38 cm x 30 cm (length x  
112 width x height) or 0.057 m<sup>3</sup> volume. This container size was selected to achieve a  
113 reasonable weight and allow manual transportation of the samples from the wetland. Prior  
114 to sampling, snow was removed from the selected sampling area in the wetland and a  
115 power saw was used to cut the frozen soil samples. Soil was excavated around the actual  
116 samples to allow insertion of straps used to lift the samples from the ground and lower  
117 them into the moving boxes. The samples were collected from an area of the treatment  
118 wetland that does not receive a high hydraulic load, in order to avoid soil samples being  
119 affected by possible accumulation of heavy metals that could lead to a high risk of  
120 pollutant leaching, as reported for the treatment peatland (Palmer et al., 2015; Khan et al.  
121 2019a).

### 122 2.1.2 Residence time and frost depth selection

123 Hydraulic residence time in the pilot wetlands was selected based on the known residence  
 124 time of the treatment wetland from which the peat material was taken, as this is a critical  
 125 hydraulic parameter controlling purification processes in treatment wetlands. According  
 126 to available data for the full-scale treatment wetland, residence time ( $T_r$ ) typically varies  
 127 from 25 to 30 days. A mean inflow rate ( $Q_i$ ) of  $2 \text{ L d}^{-1}$  to the pilot wetlands was selected.  
 128 Mean peat porosity ( $n$ ) measured at the peat sampling site was 0.9 (0.88-0.92) and thus  
 129 the mean water volume ( $V_{\text{water}} = n \cdot V$ ) in the pilot wetlands was  $0.051 \text{ m}^3$ . Using Eq. 1,  
 130 mean residence time in the pilot wetlands was determined to be 25.7 days, which was  
 131 within the range observed in the full-scale treatment peatland (frost-free period).

$$132 \quad Q_i = \frac{V_{\text{water}}}{T_r} \quad (1)$$

133 The frost depth to be achieved in the pilot wetlands was based on the sample size (depth  
 134 = 30 cm) and on frost depth data for the treatment wetland from which the peat material  
 135 was taken. The target frost depth was set between 10 and 13 cm, as this would allow for  
 136 continuous water flow underneath the frozen peat and result in a measurable effect (at  
 137 least in terms of flow regime) of freezing conditions. The freezing process was considered  
 138 to begin when the temperature dropped below  $0^\circ\text{C}$ . During frost conditions, the active  
 139 water flow volume was reduced, which affected the residence time. In determination of  
 140 water residence time in the pilot wetlands during frost conditions, the properties of the  
 141 frozen and frost-free peat layers were taken into consideration. Water volume ( $V_{\text{water}}$ )  
 142 during frost conditions was calculated using a frost-free porosity of 0.9 for the layer of  
 143 non-frozen peat and 20% of that value (i.e. 0.18) for the top frozen layer. A water content  
 144 of 20% of total porosity was selected for the frozen layer based on studies conducted by  
 145 Mustamo et al. (2019). In order to evaluate real hydraulic conditions in the pilot wetlands,  
 146 well-known tracer tests were performed under both frost-free and frost conditions (see  
 147 section 2.2.7).

### 148 2.1.3 Stabilization phase

149 Peat samples were transported to the laboratory at the University of Oulu along with about  
 150 700 L of mining-affected water (collected from the inflow to the treatment wetland). Soil  
 151 samples and mining-affected water were stored in a refrigerator ( $5^\circ\text{C}$ ) prior to the start of  
 152 the stabilisation phase. The plastic boxes were fitted with quartz sand (particle size 2-3

153 mm) compartments at the beginning and end of their longitudinal profile (width = 32 cm,  
154 length = 4 cm, height = 30 cm) and with inflow and outflow pipes (Fig. 1). Mining-  
155 affected water was introduced to the individual pilot wetlands at a rate of 2 L d<sup>-1</sup>. Electric  
156 conductivity (EC) sensors (HOBO-logger) were installed in the outflow pipes for  
157 continuous monitoring. Inflow and outflow water samples were collected at 2- to 4-week  
158 intervals and analysed for a number of contaminant concentrations (e.g. arsenic (As),  
159 antimony (Sb), dissolved organic carbon (DOC) etc.). The stabilisation phase lasted  
160 around six months (May-November 2017), although stable contaminant concentrations  
161 were observed in the outflow of both pilot wetlands after about four months (data not  
162 presented).

#### 163 *2.1.4 Preliminary freezing tests and selection of system temperature*

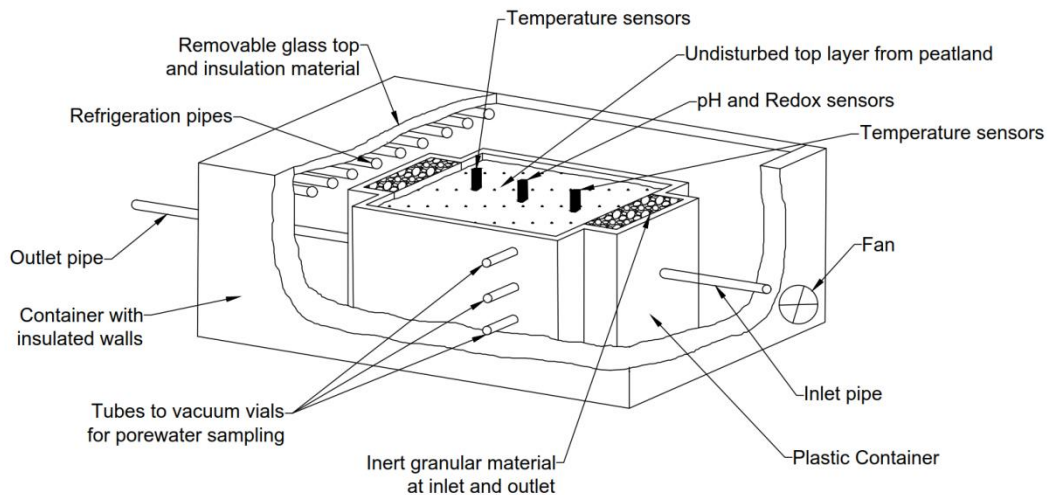
164 Preliminary freezing experiments were conducted using disturbed natural peat soil  
165 samples of the same volume, placed inside the same type of plastic boxes as were used  
166 for the actual pilot wetlands. Different refrigeration set-ups were tested and copper  
167 refrigerant tubes attached to a refrigeration unit (Lauda RK 20) were selected. Freezing  
168 was conducted from the top down and the sides of the pilot wetlands were insulated with  
169 mineral wool, to simulate *in situ* conditions as closely as possible. The refrigerant tubes  
170 were placed a couple of cm from the soil surface, in a typical configuration with bends  
171 and parallel lines. Suitable freezing temperatures were identified from trial freeze/thaw  
172 cycles. For clear, progressive frost development, an initial temperature of -20°C was  
173 found to be suitable. This was then gradually adjusted as the frost proceeded downwards  
174 from the soil surface, to keep the maximum frost depth at around 13 cm.

## 175 **2.2 Monitoring and final set-up**

### 176 *2.2.1 Sensors and their placement*

177 Temperature sensors (in-house built thermocouples attached to a logger) were installed  
178 at the soil surface and at 2, 5, 10, 13, 15, 17 and 25 cm depth, longitudinally at 10 cm  
179 from the inlet and 10 cm from the outlet of the pilot wetlands. These were used to monitor  
180 frost depth. The freezing process was considered to have started when the temperature  
181 dropped below 0°C. Redox potential sensors (Ecotech GmbH, platinum redox electrode  
182 with Ag/AgCl reference electrode, art. no. 461) and pH sensors (Ecotech GmbH, glass  
183 pH electrode with Ag/AgCl reference electrode, art. no. 465) were installed in the centre

184 of the pilot wetlands at 10 and 20 cm depth. Electric conductivity (EC) sensors (HOBO-  
 185 logger) were installed at the outflow pipes for continuous monitoring (Fig. 1, Fig. S1  
 186 (scale-drawing) in Supplementary Material).



187

188 Figure 1 – Schematic drawing of pilot wetland showing the different compartments and  
 189 placement of sensors, inflow and outflow pipes, porewater sampling tubes and  
 190 refrigeration units.

### 191 2.2.2 Porewater collection

192 Porewater samplers were installed at three depths in the pilot wetlands while they were  
 193 still in the unfrozen state. Rhizon samplers (Rhizosphere Research Products, The  
 194 Netherlands) are thin tubes made of a hydrophilic porous polymer that acts as a  
 195 microfiltration membrane. The samplers function by sucking porewater from soils and  
 196 sediments due to a negative pressure created by attaching them to an evacuated tube or a  
 197 syringe. For installation in the pilot wetlands, holes were drilled into the side of the plastic  
 198 containers (8 cm from inlet) at three different depths (5, 15 and 25 cm). Rhizon flex  
 199 samplers (pore size 0.15  $\mu\text{m}$ , filter length 5 cm, tube length 30 cm) were covered with an  
 200 outer perforated tube and inserted horizontally into the peat soil through the drilled holes  
 201 (Fig. S1 in Supplementary Material).

202 Porewater sampling was conducted daily at all depths until the frost depth reached the  
 203 layer in which the sampler was installed, leading to freezing of the porewater. That  
 204 sampler was then passed over until the layer thawed again, but samples were still taken  
 205 from unfrozen lower layers. Porewater samples were taken by attaching evacuated 12-

206 mL Exetainer® vials (Labco, UK) to the Rhizon samplers and waiting until a sufficient  
207 amount of porewater had been collected (3-5 mL). The use of evacuated vials prevented  
208 oxidation of sensitive chemical species. Samples were transferred from the vials to  
209 cryotubes using a syringe, frozen immediately in liquid nitrogen and stored at -20°C prior  
210 to analysis. In cases where the compounds of interest are not sensitive to oxygen, samples  
211 can be taken with a syringe and do not have to be kept frozen. The porewater samples  
212 were analysed for arsenic speciation (data not presented). However, porewater can be  
213 analysed for any compound of interest such as metals, total organic carbon (TOC),  
214 nitrogen compounds or sulphate.

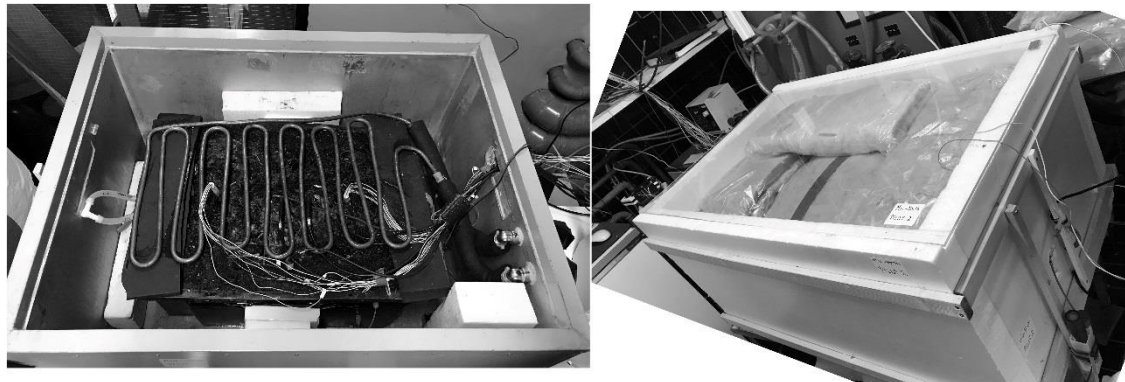
### 215 2.2.3 Gas flux sampling

216 Measuring fluxes of gases such as nitrous oxide (N<sub>2</sub>O), methane (CH<sub>4</sub>), carbon dioxide  
217 (CO<sub>2</sub>), volatile organic compounds and hydrogen sulphide (H<sub>2</sub>S) can give valuable  
218 information on the processes occurring in the soil. Gas fluxes from the pilot wetlands  
219 were measured with the static chamber method, which has been widely used to measure  
220 emissions from soils (Robertson et al., 2000; Maljanen et al., 2001). Measurements were  
221 conducted twice during the frost-free period, once directly after the onset of freezing  
222 conditions, once (first freeze-thaw cycle)/twice (second cycle) while the peat was  
223 completely frozen and once during thawing. A high-density polyethylene (HDPE)  
224 chamber (55 cm x 35 cm x 12 cm) was fitted with an airtight seal to the pilot wetlands for  
225 the period of flux measurement (Fig. S2 in Supplementary Material). As the surface area  
226 of the pilot wetland was covered completely by the chamber, the combined fluxes from  
227 the whole pilot wetland were assessed. Gas samples (~30 mL) were taken from the  
228 chamber headspace with a polypropylene syringe 5, 10, 15, 20, and 25 min after  
229 positioning of the chamber. These samples were immediately transferred to evacuate 12-  
230 mL Exetainer® vials (Labco, UK). In the present analysis we focused on CO<sub>2</sub> fluxes, as  
231 CO<sub>2</sub> production can be used as an indicator of general microbial activity in the soil. The  
232 CO<sub>2</sub> concentration in gas samples in the vials was determined using a gas chromatograph  
233 (Agilent 6890N, Agilent Technologies, USA) equipped with an autosampler (Gilson,  
234 USA) and thermal conductivity detector. Compressed air containing 398 µL/L CO<sub>2</sub> was  
235 used for calibration. Gas flux rates were calculated from the linear increase in chamber  
236 headspace concentration over time (e.g. Liikanen et al., 2006).



#### 237 2.2.4 Pilot wetland insulation and final set-up

238 The pilot wetlands were placed inside climate control units that were identical in size (80  
239 cm x 60 cm x 50 cm (length x width x height)) and were insulated in the same way.  
240 However, the unit which housed pilot wetland 1 was a pre-existing metal box and the unit  
241 which housed pilot wetland 2 was purpose-built from acrylic and polyurethane sheets and  
242 wood. To prevent freezing from the sides of the peat, the space between the wetland boxes  
243 and the walls of the host climate control unit was filled with mineral wool and fibreglass  
244 material packed between polyurethane sheets for easy application and removal. Sections  
245 of inflow and outflow pipes located inside the climate control units were also insulated to  
246 prevent freezing. In addition, insulation material was placed on the surface of the inflow  
247 and outflow compartments of the pilot wetlands to prevent freezing and blocking of  
248 inflow and outflow pipes. Thus, only the surface of the peat material was exposed to the  
249 freezing coil, which was placed about 1 cm above the soil surface. As the temperature  
250 was set to minus degrees Celsius, an ice layer formed around the coil. Insulation material  
251 (mineral wool and fibreglass packs) was also placed on top of the freezing coil (Fig. 2,  
252 Fig. S3 in Supplementary Material).



253

254 Figure 2 - Climate control units. Left: Position of the freezing coil and (right) the final  
255 set-up of the units covered with insulation material (Photos: Elisangela Heiderscheidt).

#### 256 2.2.5 Freezing/thawing cycles

257 Two freeze-thaw cycles were applied to both pilot wetlands. The first cycle lasted five  
258 weeks (Dec. 2017-Jan. 2018), while the second cycle lasted about seven weeks (Feb.  
259 2018-Mar. 2018). The frozen period was considered to begin when the temperature  
260 dropped below 0°C in the 10-13 cm soil layer. Although very similar conditions were

261 applied to both pilot wetlands, freezing and thawing of the peat progressed at slightly  
262 different rates (Table 1).

263 Table 1 – Duration of different phases of freezing and thawing in pilot wetlands 1 and 2  
264 during the first and second freeze-thaw cycles

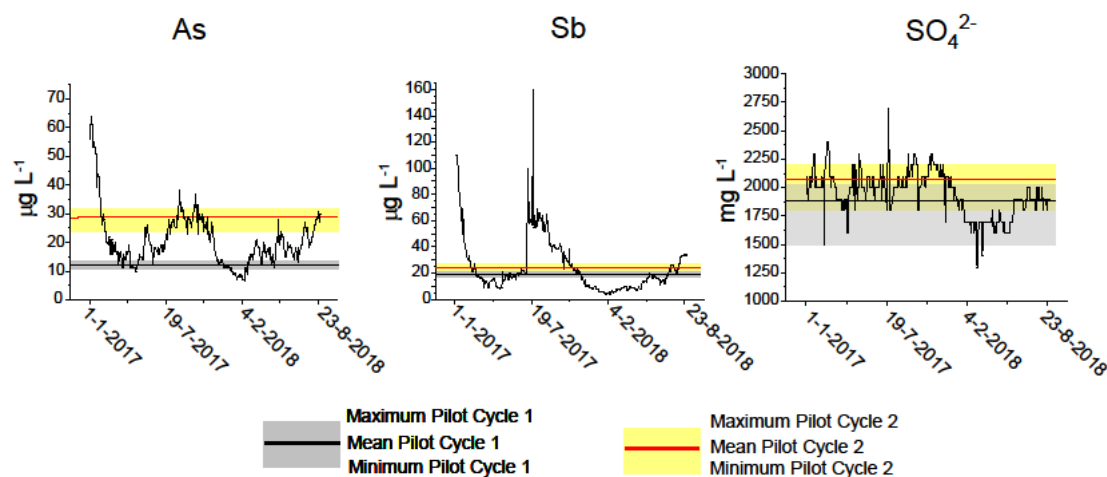
First cycle	Freezing (weeks)	Frozen (weeks)	Thawing (weeks)	Frost-free (weeks)
Pilot 1	1	2	1	1
Pilot 2	1.5	1.5	1	1
Second cycle	Freezing (weeks)	Frozen (weeks)	Thawing (weeks)	Frost free (weeks)
Pilot 1	1	4	1	1
Pilot 2	1.5	4.5	1	1

265 During the first freeze-thaw cycle, as frost developed, it caused expansion of the top  
266 layers of the wetland soil, exerting inwards pressure on the lower portion of the plastic  
267 boxes holding the soil. This caused a crack to appear at the bottom of the box in pilot  
268 wetland 1. The leak was identified and the crack was mended with multi-purpose glue  
269 and a rubber-like material. The data gathered during this period were not used in the  
270 assessment of pilot wetland 1.

#### 271 2.2.6 Water feeding and sampling

272 Inflow water (container) was kept in a refrigerator (5°C) and pumped to the pilot wetlands  
273 at an inflow rate of between 1 and 2 L d<sup>-1</sup> (frozen/frost-free). Inflow (Table S1 in  
274 Supplementary Material) and outflow water samples were collected twice per week (once  
275 in holiday periods). The samples were divided into two sub-samples, one of which was  
276 filtered (syringe filtration, GF 0.45 µm) while the other was not. Filtered and unfiltered  
277 samples were sent to accredited laboratories for analysis of dissolved and total  
278 concentrations of pollutants (e.g. SO<sub>4</sub><sup>2-</sup>, As, Sb etc.) using SFS-EN ISO methods.

279 Inflow water quality varied between the two freeze/thaw cycles as it represented different  
280 sampling periods at the full-scale treatment wetland inlet. The collected samples  
281 contained typical contaminant concentrations found in the inflow of the full-scale  
282 treatment wetland according to monitoring data for 2017 and 2018 (Fig. 3, Table S1).



283

284 Figure 3 - Inflow water concentrations of arsenic (As), antimony (Sb) and sulphate (SO<sub>4</sub><sup>2-</sup>  
 285 ) over time in the full-scale treatment wetland and quality of samples used during pilot  
 286 experiments.

### 287 2.2.7 Tracer tests for water residence time determination

288 Nominal water residence time ( $T_r$ ) in the pilot wetlands was estimated based on the water  
 289 volume ( $V_{\text{water}}$ ) of soil samples and the inflow rate ( $Q_i$ ) using Eq. (1). In order to determine  
 290 the hydraulic conditions during pilot experiments, an additional freeze/thaw cycle was  
 291 conducted and conventional tracer tests were performed. Water residence time, for both  
 292 frost-free and frozen conditions, was determined using the conservative tracer method  
 293 with sodium chloride (NaCl) (described in detail e.g. in Postila et al., 2015b). A solution  
 294 was made by diluting 254 g of NaCl in 750 mL mining-affected water. The solution was  
 295 then fed to the pilot wetlands (inflow period ~8.5 h) and EC was measured continuously  
 296 in outflow water (once per hour). Outflow tracer concentration was determined using  
 297 linear regression analysis between NaCl concentration and the resulting EC response. The  
 298 water residence time in the pilot wetlands was calculated based on the tracer response  
 299 curve and the first moment method (Kadlec and Wallace, 2009).

300 During tracer tests, inflow rate to the pilot wetlands was 2 L d<sup>-1</sup> during the first 72 days  
 301 from the time of tracer addition but was then decreased to on average 1.5 L d<sup>-1</sup> and 1.6 L  
 302 d<sup>-1</sup> for pilot wetland 1 and 2, respectively (frost-free condition). Mean inflow rate in the  
 303 tracer test under frozen conditions was 1.7 L d<sup>-1</sup> for both pilot wetlands. Normalised  
 304 residence time response curves were plotted so that hydraulic conditions in the pilot  
 305 wetlands could be compared in frost-free and frozen conditions. The outflow tracer

306 concentrations were normalised based on estimated initial concentration and time was  
307 normalised based on flow-weighted time (e.g. Werner and Kadlec, 1996).

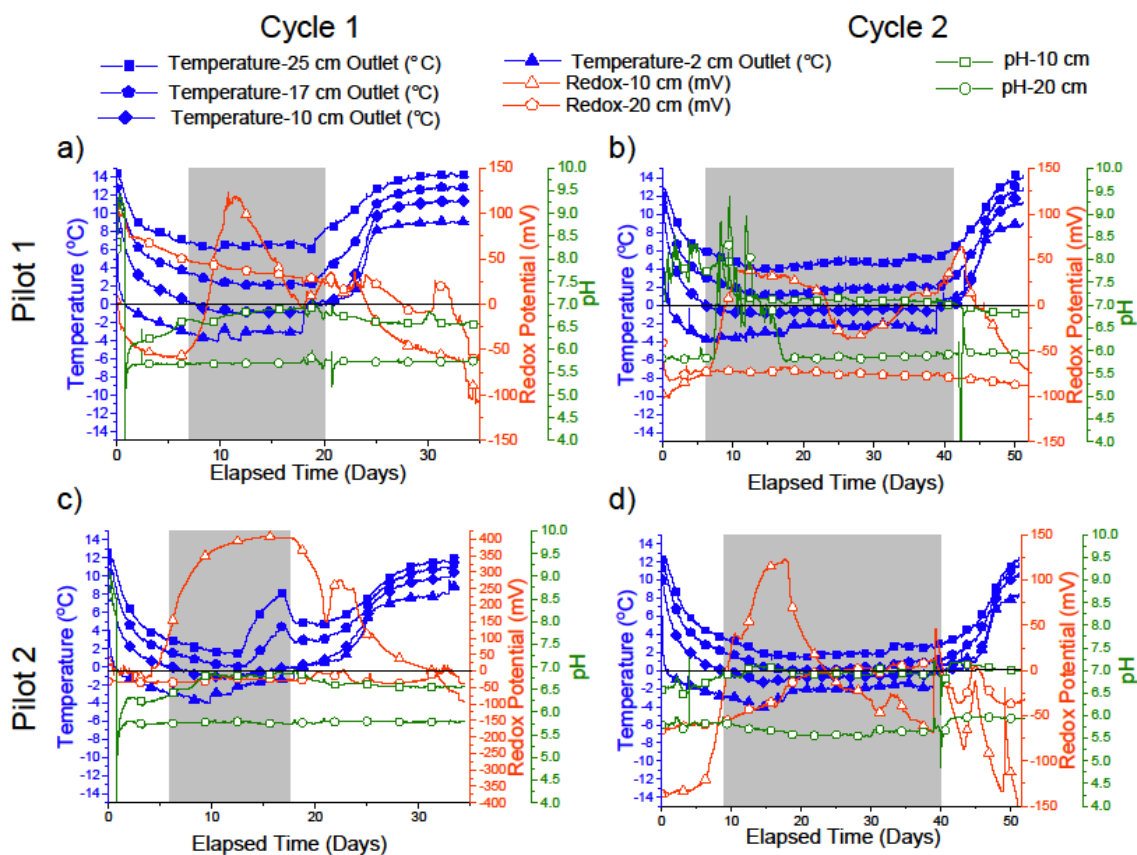
### 308 **3. Pilot wetland operation and purification results**

#### 309 **3.1 Overall functioning of pilot wetlands and monitoring equipment**

310 Two freeze/thaw cycles were applied to both pilot wetlands. Freezing of wetland peat  
311 material was carried out slowly, to mimic real conditions and avoid damage to the peat  
312 and holding containers. For example, during the first cycle, it took around 10 days for soil  
313 at 10 cm depth to reach  $-0^{\circ}\text{C}$  (temperature at 2 cm depth was  $-3^{\circ}\text{C}$ ). Peat soils have been  
314 found to freeze at temperatures lower than  $-2^{\circ}\text{C}$  and may not be fully frozen even at  $-5^{\circ}\text{C}$   
315 (Konovalov and Roman, 1973; Smerdon and Mendoza, 2010). Thus, in the pilot  
316 experiments, for temperatures higher than  $-5^{\circ}\text{C}$  partly frozen conditions can be assumed,  
317 which represent conditions commonly found in real treatment wetlands.

318 Most sensors performed as expected and data were recorded throughout the tests. The  
319 redox potential measured at 10 cm depth drastically increased following the onset of soil  
320 frost (Fig. 4). This sudden increase in redox potential was most likely due to i) pore water  
321 space around the sensor becoming smaller and its water quality changing during frozen  
322 periods and ii) increased diffusion of oxygen into the soil through cracks and channels  
323 formed by frost action (Fig. 4). The placement of the temperature sensors allowed for  
324 good monitoring of frost depth, while the data gathered by the pH, redox (20 cm) and EC  
325 sensors provided useful information regarding the processes occurring (Fig. 4). The  
326 methodology and equipment used for porewater and gas sampling proved to be sufficient,  
327 as samples were extracted and analysed without incident.

328 In future studies, additional sensors could be used to improve understanding of processes  
329 within the wetland soil. For example, oxygen sensors at different depths might give  
330 further insights on the reasons for the redox potential increase and help identify the  
331 oxic/anoxic interphase in different stages of freeze-thaw cycles. Moreover, additional  
332 porewater samplers (at more depths or at different distances from the inlet) could be used  
333 to assess e.g. the transformation of certain compounds with depth and/or distance  
334 travelled within the pilot wetland.



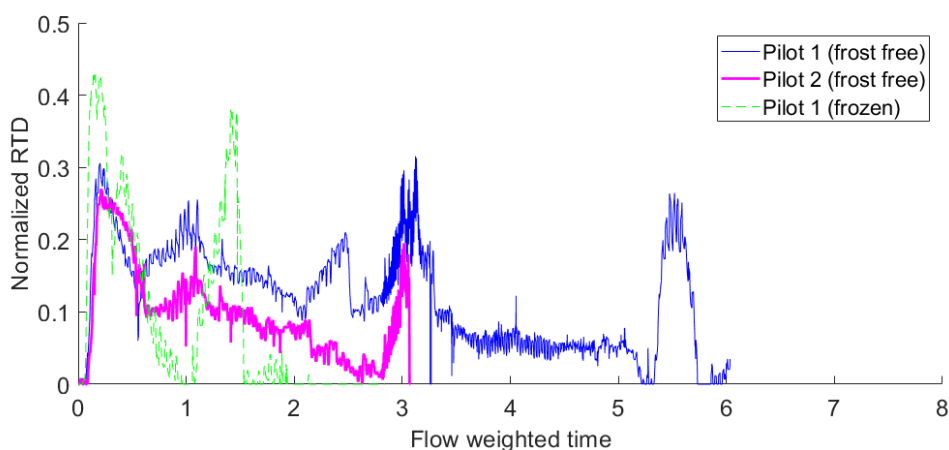
335

336 Figure 4 - Temperature (2, 10, 17 and 25 cm depth), redox and pH (10 and 20 cm depth)  
 337 measured 10 cm from the outlet of the pilot wetlands during (left) the first freeze-thaw  
 338 cycle and (right) the second freeze-thaw cycle. The shaded area indicates the frost period,  
 339 when the temperature at 10 cm soil depth remained at or below 0°C.

### 340 3.2 Hydraulic conditions

341 Based on the results of the tracer experiments, theoretically estimated water residence  
 342 time for frost-free conditions (Eq. 1) was 25.7 days for an inflow rate of 2 L d<sup>-1</sup>. During  
 343 tracer experiments, the actual inflow rate to pilot wetland 1 was 1.5 L d<sup>-1</sup> and to pilot 2  
 344 was 1.6 L d<sup>-1</sup>, leading to theoretical residence times of 34 and 32 d, respectively. Based  
 345 on tracer (NaCl) break-through curves, mean water residence time in the pilot wetlands  
 346 was much longer, 84 d (yield of tracer 71%) and 43 d (yield of tracer 30%) for pilot  
 347 wetland 1 and 2, respectively. This difference in measured mean residence time can be  
 348 due to different flow pathways within the peat, as also suggested by the five separate  
 349 peaks observed in the normalised residence time distribution curve for pilot wetland 1  
 350 (Fig. 5). The last peak occurred at around 193 days after tracer addition.

351 During frost period tracer tests, an inflow rate of  $1.7 \text{ L d}^{-1}$  was maintained and theoretical  
 352 residence time (Eq. 1) was 20 d. Mean water residence time based on the tracer break-  
 353 through curve was determined only for pilot wetland 1 and was found to be 14 d. Fewer  
 354 peaks were observed in the normalised residence time distribution curve (Fig. 5).  
 355 However, rather low yield of tracer (26%) was achieved, which can partly explain the  
 356 difference between the theoretical and tracer-based estimated mean residence time. The  
 357 lower yield of tracer observed in the test carried out during frozen compared with frost-  
 358 free conditions may be due to infiltration and retention of water (and thus NaCl) into the  
 359 frozen layer of soil (Iwata and Hirota, 2005). Generally, mean residence time was shorter  
 360 during frost than frost-free conditions, reflecting the fact that larger peat surface area and  
 361 longer reaction times are available during frost-free periods.



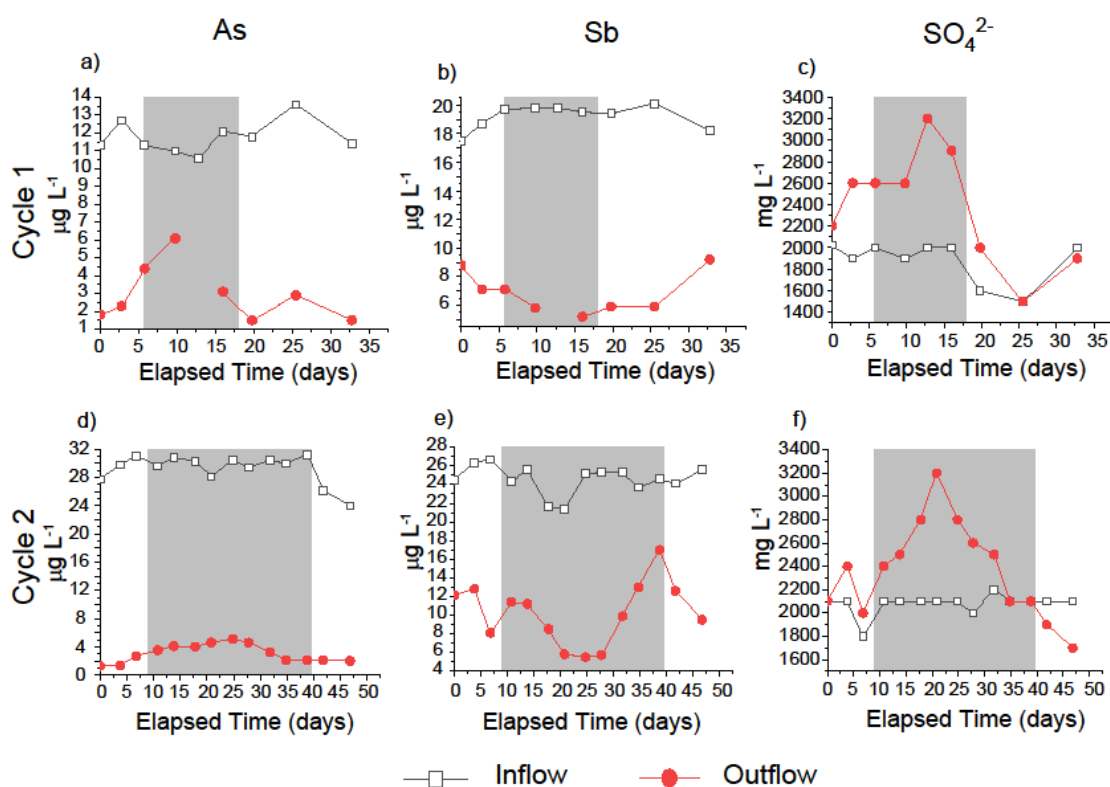
362

363 Figure 5 – Normalised tracer breakthrough curves for pilot wetlands 1 and 2. Outflow  
 364 tracer concentrations were normalised based on estimated initial concentration and time  
 365 was normalised based on flow-weighted time.

### 366 3.3 Effect of freezing and thawing on pollutant removal efficiency

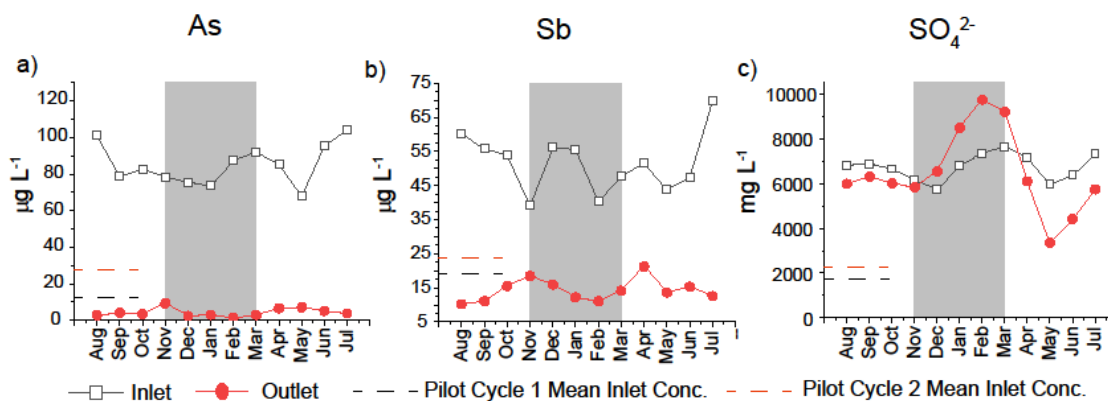
367 Variations in pollutant concentrations in outflow samples collected from the two replicate  
 368 pilot wetland systems displayed similar patterns. Thus, the removal of targeted  
 369 contaminants achieved in the two systems was comparable and the two pilot units can be  
 370 considered good replicates. Changes in purification efficiency due to freezing and  
 371 thawing conditions occurred, with higher/lower removal observed under different  
 372 conditions for different pollutants (Fig. 6). Monitoring data from the full-scale treatment  
 373 wetland were used to assess the representativeness of the results obtained (Fig. 7).  
 374 However, as sampling intervals and length of freeze/thaw cycles were much longer in the

375 full-scale system, straight comparisons between pollutant concentrations in samples from  
 376 the full-scale and pilot wetlands was not considered a viable approach. Thus a general  
 377 comparison was made focusing on seasonal variations in pollutant concentrations and  
 378 removal efficiency. Similar patterns in removal efficiency fluctuations during frost and  
 379 frost-free periods were observed in the pilot wetlands (especially for the second freeze-  
 380 thaw cycle) and in the real wetland purifying the same type of water (Fig. 7), with the  
 381 exception of a few specific compounds. Based on this general comparison, it was  
 382 concluded that the pilot wetlands simulated the full-scale system rather well.



383

384 Figure 6 – Mean concentrations of arsenic (As), antimony (Sb) and sulphate ( $SO_4^{2-}$ ) over  
 385 time in inflow and outflow from pilot wetland 2 during (upper diagrams) the first freeze-  
 386 thaw and (lower diagrams) the second freeze-thaw cycle. The shaded area indicates the  
 387 frost period, when temperature at 10 cm soil depth remained below 0 °C.

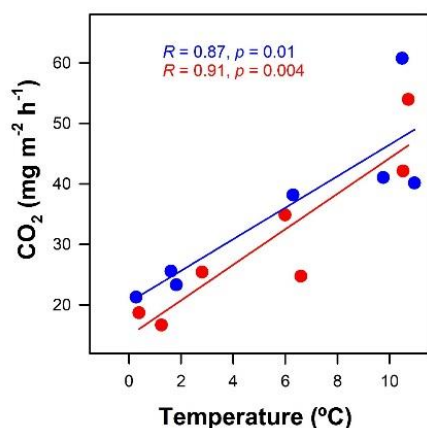


388

389 Figure 7 - Mean concentrations of arsenic (As), antimony (Sb) and sulphate (SO<sub>4</sub><sup>2-</sup>) over  
 390 time in inlet and outlet samples collected from a full-scale treatment wetland in Northern  
 391 Finland. Data are grouped by months and represent water samples collected and analysed  
 392 in the period 2010-2017.

393 Regarding the purification efficiency achieved in the pilot wetlands, it can be stated that  
 394 soil frost formation and other winter-related conditions lowered the biological process-  
 395 based removal of some pollutants (e.g. SO<sub>4</sub><sup>2-</sup>) (Fig. 6). Carbon dioxide emissions  
 396 decreased with decreasing temperature in the pilot wetlands, indicating reduced overall  
 397 biological activity (Fig. 8). However, CO<sub>2</sub> emissions were detected even when the upper  
 398 layers of the wetland soil was frozen (Fig. 8), indicating i) ongoing biological activity at  
 399 low temperatures and ii) existence of pores and channels in the frozen layers through  
 400 which gases can leave the system. As expected, flow conditions (such as residence time  
 401 of water and active water volume) changed with the development of soil frost, leading to  
 402 shorter residence time and smaller peat surface area available for pollutant retention  
 403 processes. These changes had a direct effect on pollutant removal (Fig. 6), as processes  
 404 such as sorption, biodegradation and precipitation are dependent on contact time and  
 405 media surface area, and on temperature, redox conditions and oxygen availability  
 406 (Schimel and Clein, 1996; Wang et al., 2017).





407

408 Figure 8 - Effect of peat temperature on carbon dioxide (CO<sub>2</sub>) emissions from pilot  
 409 wetlands. The average temperature for all layers (i.e. from all temperature sensors) was  
 410 used. (A good correlation was also observed between CO<sub>2</sub> emissions and the temperature  
 411 in individual peat layers; see Table S2 in Supplementary Material.)

412 Removal of arsenic decreased in the beginning of the frost periods (Fig. 6a and 6d) and  
 413 then gradually increased again to pre-frost levels. The decrease was more abrupt in the  
 414 shorter first freeze-thaw cycle than in the second cycle. Overall, redox conditions  
 415 remained mildly reducing (between -150 and +125 mV) during both freeze/thaw cycles  
 416 (Fig. 4). However, a spike was observed in redox potential at 10 cm depth in the beginning  
 417 of the frost period in both pilot wetlands during both cycles. This spike was stronger (>  
 418 +400 mV) and persisted for a longer time in pilot wetland 2 during the first freeze-thaw  
 419 cycle. The pH varied from around neutral in the surface layers (at 10 cm depth) to slightly  
 420 acidic (pH 5.5-6) in the deeper layers (at 20 cm depth). In the inflow, arsenic was mainly  
 421 in the form of arsenate (i.e. in an oxidised state; data not shown). Under the pH and redox  
 422 conditions observed in the pilot wetlands, arsenate can be reduced to arsenite by microbial  
 423 activity (Kujala et al., in review). Indeed, mainly arsenite (along with methylated As  
 424 species) was detected in the outflow (data not shown). Direct binding of arsenite via  
 425 sulfhydryl groups to natural organic matter (NOM) has been identified as a major arsenic  
 426 sequestration mechanism under reducing conditions in wetlands (Hoffmann et al., 2012;  
 427 Langner et al., 2012; Besold et al., 2018), and might also be responsible for arsenic  
 428 removal in the pilot wetlands.

429 The removal of antimony, on the other hand, increased in the beginning of the freezing  
 430 period and then gradually declined (Fig. 6b and 6e). Field data for the full-scale treatment  
 431 wetland did not indicate a significant difference in outlet antimony concentrations

432 between summer and winter months (Fig. 7b). However, it should be noted that the  
433 dilution effect of spring snowmelt and rainfall on outlet antimony concentrations reported  
434 for full-scale wetlands (Khan et al., 2019b) was not replicated in this study. Under anoxic  
435 conditions and mildly acidic pH, antimony has been shown to be sequestered by  
436 carboxyl/phenol and thiol groups in peat NOM (Besold et al., 2019). Other processes for  
437 antimony removal in wetlands include adsorption (e.g. on iron (Fe) and manganese (Mn)  
438 oxides) and precipitation (e.g. with sulphide) (Bennett et al., 2017). Exactly how frost  
439 period affected these processes to enhance antimony removal in this study is unclear.

440 The differences in removal of arsenic and antimony (Fig. 6) indicate that the pilot systems  
441 effectively replicated the diverse biogeochemical processes and pollutant  
442 competition/interactions of the full-scale mining water treatment wetland, as seen in Fig.  
443 7. However, there were some differences between the pilot wetland and field  
444 measurements, due to various reasons. Freezing of the top layer can result in surface flow  
445 in the full-scale wetland, but was not present in the pilot wetlands. Another important  
446 difference between the full-scale and pilot wetlands is the degree of contaminant  
447 accumulation in the peat. In the full-scale wetland, gradual contaminant accumulation has  
448 been detected in peat samples collected from within preferential flow areas (Khan et al.  
449 2019b). Note that the undisturbed soil samples used in the pilot wetlands were not  
450 collected from these preferential flow area. Recent outflow samples collected from the  
451 full-scale wetland actually show re-mobilisation of some of these accumulated  
452 contaminants (Khan et al. 2019b).

#### 453 **4. Conclusions**

454 Two replicate pilot wetlands were constructed using peat material from a full-scale  
455 treatment wetland and submitted to two fully controlled freeze-thaw cycles. The pilot  
456 systems were simple and constructed with low-cost materials, and were thus cost-  
457 effective. The freeze-thaw cycles satisfactorily replicated real conditions. The parameters  
458 monitored (temperature, EC, pH and redox) enabled extrapolation/comparison of the  
459 results obtained to seasonal variations observed in the full-scale treatment wetland. In  
460 addition, the parameters monitored within the wetland soil, combined with measurements  
461 of gas flux and analysis of porewater samples, provided critical clarifications regarding  
462 the processes taking place and overall process conditions. Based on the results obtained,  
463 the following conclusions can be drawn:

- 464 - The design approach and monitoring methodology developed were successful and  
465 the pilot wetlands functioned well.
- 466 - There were changes in purification efficiency due to freeze and thawing  
467 conditions in the pilot wetlands, with higher/lower removal observed in different  
468 conditions for different pollutants. Overall, soil frost formation and other winter-  
469 related conditions lowered the removal of some pollutants, for example those  
470 based on biological processes (e.g. sulphate).
- 471 - Flow conditions (such as residence time of water and active water volume)  
472 changed with the development of soil frost, leading to shorter water residence time  
473 and smaller peat surface area available for pollutant retention processes.
- 474 - Removal of arsenic and sulphate decreased during frost periods in the pilot  
475 wetlands, while removal of antimony slightly increased.
- 476 - Similar patterns in removal efficiency fluctuations during frost and frost-free  
477 periods were observed in the pilot wetlands and in the full-scale peatland-based  
478 treatment wetland. It can thus be concluded that the pilot wetlands simulated the  
479 full-scale system rather well.

480

#### 481 **Acknowledgements**

482 This study was mainly supported by the project “Development, Evaluation and  
483 Optimization of Measures to Reduce the Impact on the Environment from Mining  
484 Activities in Northern Regions (Min-North, 2016-2018)”, funded by Interreg Nord 2014-  
485 2020 program. Additional support was provided by Maj and Tor Nessling Foundation and  
486 Maa- ja vesitekniikan tuki ry. The authors would like to acknowledge the support received  
487 from the personnel at the Water, Energy and Environmental Engineering research unit at  
488 the University of Oulu, in particular from laboratory technicians Tuomo Reinikka and  
489 Tuomo Pitkänen.

490

#### 491 **References**

492 Bennett, W.W., Hockmann, K., Johnston, S.G., Burton, E.D., 2017. Synchrotron X-ray  
493 absorption spectroscopy reveals antimony sequestration by reduced sulfur in a freshwater  
494 wetland sediment. *Environ. Chem.* 14, 345–349.

- 495 Besold, J., Biswas, A., Suess, E., Scheinost, A.C., Rossberg, A., Mikutta, C. 2018.  
496 Monothioarsenate transformation kinetics determines arsenic sequestration by sulfhydryl  
497 groups of peat. *Environ Sci Technol* 52: 7317-7326.
- 498 Besold, J., Kumar, N., Scheinost, A.C., Lezama Pacheco, J., Fendorf, S., Planer-Friedrich,  
499 B., 2019. Antimonite complexation with thiol and carboxyl/phenol groups of peat organic  
500 matter. *Environ. Sci. Technol.* 53, 5005–5015. <https://doi.org/10.1021/acs.est.9b00495>
- 501 Cleavitt, N.L., Fahey, T.J., Groffman, P.M., Hardy, J.P., Henry, K.S., Driscoll, C.T.,  
502 2008. Effects of soil freezing on fine roots in a northern hardwood forest. *Can. J. For.*  
503 *Res.* 38, 82–91
- 504 Edwards, A. C., and M. S. Cresser. 1992. Freezing and its effect on chemical and  
505 biological properties of soil. *Adv. Soil Sci.* 18:59-79.
- 506 Fitzhugh, R.D., Driscoll, C.T., Groffman, P.M., Tierney, G.L., Fahey, T.J., Hardy, J.P.,  
507 2003. Soil freezing and the acid-base chemistry of soil solutions in a northern hardwood  
508 forest. *Soil Sci. Soc. Am. J.* 67, 1897. <https://doi.org/10.2136/sssaj2003.1897>
- 509 Groffman, P.M., Hardy, J.P., Driscoll, C.T., Fahey, T.J., 2006. Snow depth, soil freezing,  
510 and fluxes of carbon dioxide, nitrous oxide and methane in a northern hardwood forest.  
511 *Glob. Chang. Biol.* 12, 1748–1760.
- 512 Heikkinen, K., Karppinen, A., Karjalainen, S.M., Postila, H., Hadzic, M., Tolkkinen,  
513 Marttila, H., Ihme, R., Kløve, B., 2018. Long-term purification efficiency and factors  
514 affecting performance in peatland-based treatment wetlands: an analysis of 28 peat  
515 extraction sites in Finland. *Ecol. Eng.* 117, 153–164.
- 516 Hoffmann, M., Mikutta, C., and Kretzschmar, R., 2012. Bisulfide reaction with natural  
517 organic matter enhances arsenite sorption: Insights from X-ray adsorption spectroscopy.  
518 *Environ Sci Technol* 46: 11788-11797.
- 519 Iwata, Y., Hirota T., 2005. Monitoring over-winter soil water dynamics in a freezing and  
520 snow-covered environment using a thermally insulated tensiometer. *Hydrological*  
521 *Processes* 19, 3013–3019.
- 522 Kadivar, M., Manahiloh, K.N., 2019. Revisiting parameters that dictate the mechanical  
523 behavior of frozen soils. *Cold Reg. Sci. Technol.*, 163, 34-43.

- 524 Kadlec, R.H., Reddy, K.R., 2001. Temperature effects in treatment wetlands. *Water*  
525 *Environ Res.*,73(5), 543-57.
- 526 Kadlec, R.H., Wallace, S.D., 2009. *Treatment Wetlands*. Second edition, CRC Press Boca  
527 Raton, Florida, USA.
- 528 Khan, U., Kujala, K., Nieminen, S., Räisänen, M., Ronkanen, A-K., 2019a. Arsenic,  
529 antimony and nickel leaching from northern peatlands treating mining influenced water  
530 in cold climate. *Science of the Total Environment*, 657, 1161-1172
- 531 Khan, U.A., Kujala, K., Planer-Friedrich, B., Räisänen, M., Ronkanen, A-K., 2019b.  
532 Seasonal and long-term trends of Sb retention and mobilization in boreal treatment  
533 peatlands. (Manuscript submitted for publication)
- 534 Knoblauch, C., Jørgensen B.B., 1999. Effect of temperature on sulphate reduction,  
535 growth rate and growth yield in five psychrophilic sulphate-reducing bacteria from Arctic  
536 sediments. *Environmental Microbiology*, 1, 457-467.
- 537 Konovalov, A., Roman, L., 1973. The thermophysical properties of peat soils. *Soil*  
538 *Mechanics and Foundation Engineering*. 10, 179-181
- 539 Kujala, K., Karlsson, T., Nieminen, S., Ronkanen, A-K., 2019. Design parameters for  
540 nitrogen removal by constructed wetlands treating mine waters and municipal wastewater  
541 under Nordic conditions. *Science of the Total Environment*, 662, 559-570.
- 542 Kujala, K., Besold, J., Mikkonen, A., Tirola, M., Planer-Friedrich, B. Abundant and  
543 diverse arsenic-metabolizing microorganisms in peatlands treating arsenic-contaminated  
544 mining wastewaters. Submitted to *Environmental Microbiology*.
- 545 Langner, P., Mikutta, C., Kretzschmar, R., 2012. Arsenic sequestration by organic  
546 sulphur in peat. *Nat. Geosci.* 5, 66–73.
- 547 Liikanen, A., Huttunen, J.T., Karjalainen, S.M., Heikkinen, K., Väisänen, T.S., Nykänen,  
548 H., Martikainen, P.J., 2006. Temporal and seasonal changes in greenhouse gas emissions  
549 from a constructed wetland purifying peat mining runoff waters. *Ecological Engineering*  
550 26:241-251.
- 551 Maljanen, M., Hytönen, J., Martikainen, P.J., 2001. Fluxes of N<sub>2</sub>O, CH<sub>4</sub> and CO<sub>2</sub> on  
552 afforested boreal agricultural soils. *Plant and Soil*, 231, 113-121.

- 553 Mustamo, P., Ronkanen, A.-K., Berglund, Ö., Berglund, K., Kløve, B., 2019. Thermal  
554 conductivity of unfrozen and partially frozen managed peat soils. *Soil Tillage Res.* 191,  
555 245–255.
- 556 Palmer K., Ronkanen A-K., Kløve B. 2015. Efficient removal of arsenic, antimony and  
557 nickel from mining wastewaters in Northern treatment peatlands and potential risks in  
558 their long-term use. *Ecological Engineering*, 75, 350-364.
- 559 Postila, H., Ronkanen, A.-K., Kløve, B., 2015a. Wintertime purification efficiency of  
560 constructed wetlands treating runoff water from peat extraction in cold climate.  
561 *Ecological. Engineering* 85:13-25.
- 562 Postila, H., Ronkanen, A.-K., Marttila, H., Kløve, B. 2015b. Hydrology and hydraulics  
563 of treatment wetlands constructed on drained peatlands. *Ecological Engineering* 75: 232–  
564 241.
- 565 Robertson, G.P., Paul, E.A., Harwood, R.R., 2000. Greenhouse gases in intensive  
566 agriculture: contributions of individual gases to the radiative forcing of the atmosphere.  
567 *Science*, 289, 1922-1925.
- 568 Ronkanen, A-K., Kløve, B. 2007. Use of stable isotopes and tracers to detect preferential  
569 flow patterns in a peatland treating municipal wastewater. *Journal of Hydrology* 347:418-  
570 429.
- 571 Schimel, J.P., Clein, J.S., 1996. Microbial response to freeze–thaw cycles in tundra and  
572 taiga soils. *Soil Biology and Biochemistry* 28, 1061–1066.
- 573 Smerdon, B.D., Mendoza, C.A., 2010. Hysteretic freezing characteristics of riparian  
574 peatlands in the Western Boreal Forest of Canada. *Hydrological Processes* 24, 1027-1038
- 575 Smid, A.E., Beauchamp, E.G., 1976. Effects of temperature and organic matter on  
576 denitrification in soil. *Canadian Journal of Soil Science* 56, 385-391.
- 577 Stein, O.R., Borden-Stewart, D.J., Hook, P.B., Jones, W.L., Seasonal influence on sulfate  
578 reduction and zinc sequestration in subsurface treatment wetlands. *Water Research* 41,  
579 3440-3448.
- 580 Sulkava, P., Huhta, V., 2003. Effects of hard frost and freeze-thaw cycles on decomposer  
581 communities and N mineralisation in boreal forest soil. *Appl. Soil Ecol.* 22, 225–239.

- 582 Tourna M., Freitag, T.E., Nicol, G.W., Prosser, J.I., 2008. Growth, activity and  
583 temperature responses of ammonia-oxidizing archaea and bacteria in soil microcosms.  
584 *Environmental Microbiology*, 10, 1357-1364.
- 585 Vymazal, J., 2011. Constructed wetlands for wastewater treatment: five decades of  
586 experience. *Environ. Sci. Technol.* 45, 61–69.
- 587 Wang, M., Zhang, D.Q., Dong, J.W., Tan, S.K., 2017. Constructed wetlands for  
588 wastewater treatment in cold climate — A review. *Journal of Environmental Sciences*,  
589 57, 293-311.
- 590 Werner, T.M., Kadlec, R.H., 1996. Application of residence time distributions to  
591 stormwater treatment systems. *Ecological Engineering* 7, 213-234.
- 592 Zhang, T., Barry, R.G., Armstrong, R.L., 2004. Application of satellite remote sensing  
593 techniques to frozen ground studies. *Polar Geogr.* 28, 163-196